γ -graphyne: A promising electron acceptor for organic photovoltaics

The most significant discovery of recent years is graphene, which consists of a single layer of sp2hybrid carbon atoms arranged in a two-dimensional (2D) honeycomb lattice nanostructure. Graphynes – a two-dimensional materials similar to graphene, but composed of sp-hybridized carbons periodically integrated into a sp²-hybridized carbon network. Structure of graphynes were first theoretically proposed in 1987 by Baughman *et al.*[1] More than 20 years later, in 2010, Li et al. developed the first successful methodology for creating γ -graphdiyne films.[2] Despite the significant attention from the scientific community, only recently Hu et al. reported the synthesis of γ -graphyne with unified long-range crystalline structure.[3]

We demonstrate that γ -graphyne is an efficient electron acceptor due to its low LUMO and its ability to delocalize an excess charge.[4,5] Moreover, in contrast to other 2D sheets like graphene its electronic properties are not sensitive to vacancy defects. Investigation of photoinduced electron transfer processes in complexes of γ -graphyne with typical electron-accepting and electrondonating partners shows that the lowest excited states with donors are charge separated states formed by the electron transfer from partner to γ -graphyne. This electron transfer is thermodynamically favorable and occurs on nano-to-picosecond time scale. In contrast, electron transfer from γ -graphyne to strong electron acceptors such as fullerenes or arylenediimides is unlikely.

References:

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